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# DIFFUSION: FROM THE ATOMIC TO THE MACROSCOPIC SCALE

### MAYLISE NASTAR

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### **TRACER DIFFUSION COEFFICIENT**

### Fick's first law

(salt diffusion in liquid, 1855)



$$J_{A^*} = -D_{A^*} \nabla C_{A^*}$$

\_\_\_\_\_

$$\frac{\partial c_{A^*}}{\partial t} = -div(J_{A^*}) = D_{A^*} \Delta c_{A^*}$$

Continuity equation

- J : nb of atoms/m<sup>2</sup>/s
- D : diffusion coefct m<sup>2</sup>/s
- C : nb atoms/m<sup>3</sup>

1D analytical solution  $C_{A^*}(x,t=0) = M\delta(x)$  $C_{A^*}(x,t) = \frac{M}{\sqrt{\pi Dt}} \exp\left(-\frac{x^2}{4Dt}\right)$ 

#### **Diffusion experiment in solid** (\*Pb tracer,Von Hevesy, 1921)





### From macroscopic to microscopic diffusion

Interdiffusion in Cu-Zn Kirkendall, 1942



Displacement of the markers during interdiffusion! Kirkendall explained it by introducing vacancies (after his thesis defense)



# Vacancy-atom exchange frequency

Transition State Theory \*H. Heyring, 1935

If t<sub>vib</sub>,t<sub>jump</sub> << t<sub>residence</sub> the exchanges can be described on a rigid lattice using a thermally activated frequency

$$W_{BV} = V_B e^{-\frac{E_{mig}}{k_B T}}$$





\*A.B. Lidiard. Phil. Mag., 46 (7) (1955)



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### A multiscale approach for diffusion and phase transformations







<u>1.Diffusion: a property that can be measured at various scales</u>
a) Onsager formalism versus Fick's laws
b) Interdiffusion within the Onsager formalism
c) Nanoscale diffusion in multilayers

2. Modeling of diffusion and diffusion data for what?

- a) Parameterization of the Atomic Kinetic Monte Carlo
- b) Lattice rate theory for driven systems
- c) Phase Field method
- d) Cluster Dynamics

### 3. Diffusion from the microscopic diffusion processes

- a) Lij from the equilibrium fluctuations
- b) Self-consistent mean field theory: generalized diffusion equation
- c) Applications of the SCMF theory

### **ONSAGER FORMALISM**





Flux couplings not reproduced by a Fick's law



# Onsager definition of fluxes

\*L. Onsager, Phys. Rev. 37, 405 (1931); 38, 2265 (1931).

$$\mu_{\mathbf{x}}^{i} = \frac{\partial F}{\partial N_{\mathbf{x}}^{i}} \bigg|_{N_{\mathbf{x}}^{j}}$$

<u>Chemical potential</u> as conjugate of the extensive conserved variable  $\left(N_{x}^{i}\right)$ 

<u>Flux of particules</u> Within the linear response theory:



$$J_{i} = -\sum_{j} L_{ij} \frac{\left(\mu_{x+1}^{j} - \mu_{x}^{j}\right)}{k_{B}T}$$

-Diffusion driving forces are gradients of chemical potentials

- -Lij are equilibrium quantities (that do not depend on the diffusion experiment) -Production of entropy: positive definite Onsager matrix (Lij)
- -Onsager-Casimir's reciprocal relation (1931):
- Microscopic time reversal (detailed balance in lattice-based diffusion) leads to symmetric Onsager matrix (Lij)

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### Onsager coefts versus Diffusion coefts

#### Fluxes in a binary alloy (ABV):

$$\begin{cases} J_{A} = -L_{AA} \nabla \mu_{A} - L_{AB} \nabla \mu_{B} - L_{AV} \nabla \mu_{V} \\ J_{B} = -L_{BA} \nabla \mu_{A} - L_{BB} \nabla \mu_{B} - L_{BV} \nabla \mu_{V} \\ J_{V} = -J_{A} - J_{B} \end{cases}$$

Three independent Onsager coefts and two driving forces

$$\begin{cases} J_{A} = -L_{AA} \nabla(\mu_{A} - \mu_{V}) - L_{AB} \nabla(\mu_{B} - \mu_{V}) \\ J_{B} = -L_{AB} \nabla(\mu_{A} - \mu_{V}) - L_{BB} \nabla(\mu_{B} - \mu_{V}) \\ L_{AV} = -L_{AA} - L_{AB} \\ L_{BV} = -L_{BB} - L_{AB} \end{cases}$$

**Diffusion coefficients** 



#### D<sub>Cu</sub> -13 D<sub>Cu</sub> .og (diffusion coefficient, m²/sec) 14 $D_{\rm Ni}^{\star} (X_{\rm Cu} = 1)$ $D_{Cu}^{\star}(X_{Ni}=1)$ $D_{\rm Ni}$ -15 $D_{Ni}^*$ -16 Cu 0.2 0.4 0.6 0.8 Ni Atomic fraction nickel

Fig. 2.21 The relationship between the various diffusion coefficients in the Cu-Ni system at 1000 °C (After A.G. Guy, *Introduction to Materials Science*, McGraw-Hill, New York, 1971.)

#### Experimental diffusion coefts

# Concept and a nightmare

Introduces a second time-scale, off diagonal  $L_{AB}$  and correlations



Vacancy: an equilibrium structural defect



Vacancy in most solid systems is a non conservative species: the vacancy number  $N_V$  is deduced from the minimization of the free enthalpy with respect to  $N_V$ 



# **Gibbs-Duheim relationship**

Extensive variables: free energy (F) number of atoms of species i ( $N_i$ )

Variation of F depends on the variation of the extensive  $N_i$  variables

Conjugate variable of  $N_i$ : intensive variable that depends on intensive variables only

Free energy F de deduced from  $F = \int_{0}^{N} \sum_{i} \mu_{i} dN_{i} = N \int_{0}^{N} \sum_{i} \mu_{i} dC_{i} = N \sum_{i} \mu_{i} C_{i} = \sum_{i} \mu_{i} N_{i}$ 

At equilibrium:

$$dF = 0 \Rightarrow dF = \sum_{i} N_{i} d\mu_{i} + \mu_{i} dN_{i} = 0 \Rightarrow \sum_{i} N_{i} d\mu_{i} = 0$$

Gibbs-Duheim relationship in alloy ABV

$$C_{A}\nabla\mu_{A} + C_{B}\nabla\mu_{B} + C_{V}\nabla\mu_{V} = 0$$

$$dF = \sum_{i} \mu_{i} dN_{i}$$

$$u_{i} = \frac{\partial F}{\partial N_{i}} \bigg|_{V,T,N_{i}}$$

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Diffusion driving forces in terms of alloy & vacancy chemical potentials

Gibbs-Duheim: 
$$C_A \nabla \mu_A + C_B \nabla \mu_B + C_V \nabla \mu_V = 0$$

$$\nabla(\mu_{A} - \mu_{V}) = -\frac{C_{B}}{C_{A} + C_{B}} \nabla(\mu_{B} - \mu_{A}) - \frac{1}{C_{A} + C_{B}} \nabla\mu_{V}$$
  
Alloy driving force vacancy driving force

Definition of the thermodynamic factor  $\Phi$ 

$$\nabla C_{V} \ll \nabla C_{B} \qquad \qquad \frac{\nabla (\mu_{B} - \mu_{A})}{k_{B}T} = \frac{\Phi}{C_{A}C_{B}} \nabla C_{B}$$

A Bragg-Williams mean field expression:  $\Phi_{BW} = 1 - 4C_B(1 - C_B)\frac{T_C}{T}$ 

### INTERDIFFUSION WITHIN THE ONSAGER FORMALISM



# Interdiffusion in binary alloys

### <u>nonconservative vacancy</u>: $\nabla \mu_v = 0$

 $\frac{\text{Diffusion driving forces}}{\text{Thermodynamic factor } \Phi}$ 

$$\nabla(\mu_A - \mu_V) = -C_B \frac{\Phi}{C_A C_B} \nabla C_B$$
$$\nabla(\mu_B - \mu_V) = C_A \frac{\Phi}{C_A C_B} \nabla C_B$$

Fluxes in the frame of the cristal lattice

Intrinsic diffusion coeft D<sub>i</sub>

Kirkendall speed v<sub>k</sub>

$$J_{A} = -\left(\frac{L_{AA}}{C_{A}} - \frac{L_{AB}}{C_{B}}\right) \Phi \nabla C_{A} = -D_{A} \nabla C_{A}$$
$$J_{B} = -D_{B} \nabla C_{B}$$

$$J_{V} = -(J_{A} + J_{B}) = V_{K}$$

Fluxes in the frame of the laboratory displacement of the lattice

$$\mathbf{J}_{\mathbf{A}}^{0} = \mathbf{J}_{\mathbf{A}} + \mathbf{C}_{\mathbf{A}}\mathbf{v}_{\mathbf{K}} = -(\mathbf{C}_{\mathbf{A}}\mathbf{D}_{\mathbf{B}} + \mathbf{C}_{\mathbf{B}}\mathbf{D}_{\mathbf{A}})\nabla\mathbf{C}_{\mathbf{A}}$$

Interdiffusion coefficient D

$$\frac{dC_A^0}{dt} = D\nabla^2 C_A^0 \quad \text{with} \quad D = C_A D_B + C_B D_A$$

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# Interdiffusion in binary alloys

$$\frac{\text{conservative vacancy}}{\nabla \mu_V \neq 0}$$

$$\begin{cases} J_{A} = +M_{CV} \frac{\nabla(\mu_{B} - \mu_{A})}{k_{B}T} - \frac{L_{BB} + L_{AB}}{L_{AA} + L_{BB} + 2L_{AB}} J_{V} \\ J_{B} = -M_{CV} \frac{\nabla(\mu_{B} - \mu_{A})}{k_{B}T} - \frac{L_{AA} + L_{AB}}{L_{AA} + L_{BB} + 2L_{AB}} J_{V} \end{cases}$$

Adiabaticity of V:  $\frac{dC_V}{dt} = 0$ 

$$\frac{\mathrm{d}\mathrm{C}_{\mathrm{A}}}{\mathrm{d}\mathrm{t}} = -\nabla\mathrm{J}_{\mathrm{A}} = \mathrm{M}_{\mathrm{CV}} \frac{\Phi}{\mathrm{C}_{\mathrm{A}}\mathrm{C}_{\mathrm{B}}} \nabla^{2}\mathrm{C}_{\mathrm{A}}$$

Mobility:  $M_{CV} = \frac{L_{AA}L_{BB} - L_{AB}^{2}}{L_{AA} + 2L_{AB} + L_{BB}}$ 

Conservative interdiffusion coefficient:  $D = M_{CV} \frac{\Phi}{C_A C_B}$ 

### Two interdiffusion mobilities

Non Conservative vacancy mobility  

$$M_{\tilde{D}} = C_{A} \left[ C_{B} L_{AA} - C_{A} L_{AB} \right] + C_{B} \left[ C_{A} L_{BB} - C_{B} L_{AB} \right]$$

Conservative vacancy mobility  
$$M_{CV} = \frac{L_{AA}L_{BB} - {L_{AB}}^2}{L_{AA} + 2L_{AB} + L_{BB}}$$

Mobility taking into account the second driving force produced by Kirkendall induced vacancies



Fig. 2.15 Interdiffusion and vacancy flow. (a) Composition profile after interdiffusion of A and B. (b) The corresponding fluxes of atoms and vacancies as a function of position x. (c) The rate at which the vacancy concentration would increase or decrease if vacancies were not created or destroyed by dislocation climb.





### A mobility that depends on the microstructure



Percolated microstructure



C<sub>B</sub>=0.15



Non percolated microstructure



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### The measurement of concentration dependent D

The Matano's method • Graphic solution  $D(C') := \frac{-\int_{C_1}^C x \, dC}{2t \left(\frac{\partial C}{\partial x}\right)_{C'}}$ • The Matano's plane (x = 0) is defined by :  $\Rightarrow \int_{C_1}^{C_2} x \, dC = 0$ 





### **Classical diffusion experiments**

Tracer A\* in A 
$$\nabla \mu_{\rm B} = 0 \nabla \mu_{\rm A*} = -(c_{\rm A} / c_{\rm A*}) \nabla \mu_{\rm A}$$
  
$$J_{A*} = -L_{A*A*} \left( \frac{1}{C_{A*}} - \frac{L_{A*A}}{C_{A}L_{A*A*}} \right) \nabla C_{A}^{*}$$

Intrinsic diffusion coefficients  $\nabla \mu_{B} = -(c_{A} / c_{B}) \nabla \mu_{A}$ 

$$\begin{cases} J_{A} = -\left(\frac{L_{AA}}{C_{A}} - \frac{L_{AB}}{C_{B}}\right) \Phi \nabla C_{A} = -D_{A} \nabla C_{A} \\ J_{B} = -D_{B} \nabla C_{B} \end{cases}$$

Interdiffusion coefficients

$$\tilde{\mathbf{D}} = \mathbf{C}_{\mathbf{B}}\mathbf{D}_{\mathbf{A}} + \mathbf{C}_{\mathbf{A}}\mathbf{D}_{\mathbf{B}}$$

Wind factor: not easily measured



 $G = \frac{L_{BV}}{L_{BB}}$ 



Fig. 2.21 The relationship between the various diffusion coefficients in the Cu-Ni system at 1000 °C (After A.G. Guy, *Introduction to Materials Science*, McGraw-Hill, New York, 1971.)



# A lack of experimental data

Partial characterization of diffusion properties in binary alloys:

-two intrinsic diffusion coefficients only for the determination of 3 Onsager coefficients -Even for steels, diffusion coefficients have been measured for a narrow range of composition and temperature (high temperature)

Some phenomenological approximations

**Darken:**  $D_i = D_i^* \Phi$ 

### CALPHAD-DICTRA:

 $L_{ii} = 0$  if  $i \neq j$ 

Manning's relations

 $L_{ij} = \frac{C_i D_i^*}{k_B T} \left| \delta_{ij} + \frac{1 - f_0}{f_0} \frac{C_j D_j^*}{\sum C_k D_k^*} \right|$  does not produce negative off-diagonal  $L_{ij}$ 

Can we extrapolate low-T diffusion properties from high T? No data for diffusion mechanisms ocurring under irradiation only (split interstitial) Is the possible change of sign of the off-diagonal Lij affecting kinetics?

### NANOSCALE DIFFUSION IN MULTILAYERS



FIG. 5. The effective diffusion coefficient  $\widetilde{D}_{\lambda}$  as a function of  $B^2$  at 200, 225, and 260 °C.

 $k^2 (10^{14} cm^{-2})$ 

# Free energy of an inhomogeneous system

 $C_B^{\beta}$  $C^{\alpha}_{p}$ 

$$\left(12^{2}\right)^{2}$$

\*Cahn and Hilliard, 1958

$$F = AN_V \int_L \left( f + \frac{1}{2} \kappa \left( \frac{dC_B}{dx} \right)^2 \right) dx$$

A: surface and V: total volum V=A\*L  $N_V$ : nbr of atoms per unit volum

A phase field kinetic equation (\*Cahn, 1965)  $\frac{dC_B}{dt} = M\nabla^2 \frac{\delta F}{\delta C_B(x)} = M\nabla^2 \left(\frac{\partial f}{\partial C_B} - \kappa \frac{d^2 C_B}{dx^2}\right) \quad M = \frac{D}{\partial^2 f / \partial C_B^2}$ 

It is assumed that a similar expression of the free energy can be used in non equilibrium systems and the mobility is related to the interdiffusion coeft D



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Expansion of the alloy chemical potential around a homogeneous solid solution

$$\mu_{B} - \mu_{A} = \left(\mu_{B} - \mu_{A}\right)_{k=0} + \left(\frac{\partial^{2} f}{\partial C_{B}^{2}} + \kappa k^{2}\right) \delta C_{B}$$



 + sinusoïdal composition fluctuation with wavevector k



Composition gradient parameter  $\kappa$  deduced from the equilibrium composition fluctuations

$$S(k) = \delta C_B(k) \delta C_B(-k) = \frac{1}{\frac{\partial^2 f}{\partial C_B^2} + \kappa k^2}$$

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#### THE CAHN-HILLIARD LINEAR STABILITY ANALYSIS (1958-65)









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#### ATOMIC KINETIC MONTE CARLO SIMULATIONS OF NANOSCALE DIFFUSION IN FE-CR



The composition-gradient parameter depends on the diffusion mechanism!



## OUTLINE

1.Diffusion: a property that can be measured at various scales
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<u>2. Modeling of diffusion and diffusion data for what?</u>
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### PARAMETERIZATION OF THE AKMC SIMULATIONS

# Monte Carlo simulation with a Residence time algorithm

### Activated jump frequency



Broken bond model of the activation energy

$$E_{A}^{act} = E_{A}^{saddle \, point} - \underbrace{\sum V_{Aj} - \sum V_{iV}}_{broken \, bonds}$$

Master Equation of a Markov process

$$\frac{\mathrm{dP}}{\mathrm{dt}} = \sum_{l\neq n} B_{nl} P_l(t) - B_{ln} P_n(t)$$



Residence time algorithm



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### PHYSICAL TIME SCALE AND POINT DEFECT CONCENTRATIONS IN MONTE CARLO SIMULATIONS

 $\langle C_{V}^{MC} \rangle = 1 / N$  in a Monte Carlo box versus  $\langle C_{V}^{eq} \rangle$  in an alloy

If diffusion coefficient linear with Cv then

time rescaling: 
$$t = t_{MC} \langle C_V^{MC} \rangle / \langle C_V^{eq} \rangle$$
 (volves with time

If vacancy is at equilibrium then for any local environment  $\boldsymbol{\alpha}$ 

$$t = t_{MC} \frac{C_V^{MC}(\alpha)}{C_V^{eq}(\alpha)}$$

possible reference environments: pure A and pure B

Estimation of total and local vacancy concentrations:

$$\left\langle C_{V}^{eq} \right\rangle = \left\langle C_{V}^{MC} \right\rangle \frac{C_{V}^{eq}(\alpha)}{C_{V}^{MC}(\alpha)} \qquad \left\langle C_{i}^{V} \right\rangle_{\tau} = \left\langle C_{V}^{eq} \right\rangle \frac{\tau_{i}^{V}}{\tau}$$

\*Y. Le Bouar, F. Soisson, PRB 2007, M. Nastar&F.Soisson, PRB 2012

### LATTICE RATE THEORY FOR DRIVEN SYSTEMS

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### Radiation Induced Segregation (RIS) in steels



\*Busby, Was, Bruemmer, Edwards, Kenik, MRS Symp. Proc., 1999, **540**, 451

\*Wharry, Was , JNM 442 (2013)





RIS at the atomic scale



3
# Fluxes in binary alloys

$$\begin{cases} J_A^V = -L_{AA}^V \nabla \mu_A - L_{AB}^V \nabla \mu_B - L_{AV} \nabla \mu_V \\ J_B^V = -L_{AB}^V \nabla \mu_A - L_{BB}^V \nabla \mu_B - L_{BV} \nabla \mu_V \\ J_V = -(J_A^V + J_B^V) \end{cases}$$
$$\begin{cases} J_A^I = -L_{AA}^I \nabla \mu_A - L_{AB}^I \nabla \mu_B - L_{AI} \nabla \mu_I \\ J_B^I = -L_{AB}^I \nabla \mu_A - L_{BB}^I \nabla \mu_B - L_{BI} \nabla \mu_I \\ J_I = +(J_A^I + J_B^I) \end{cases}$$

2x3 independant Lij coefcts

$$\begin{cases} L_{AV} = -(L_{AA}^{V} + L_{AB}^{V}) \\ L_{BV} = -(L_{BB}^{V} + L_{AB}^{V}) \\ L_{AI} = +(L_{AA}^{I} + L_{AB}^{I}) \\ L_{BI} = +(L_{BB}^{I} + L_{AB}^{I}) \end{cases}$$

### Driving forces in binary alloys

Chemical driving force

$$\frac{\nabla \mu_{\rm A}}{k_{\rm B}T} = \frac{\Phi}{C_{\rm A}} \nabla C_{\rm A} \quad \frac{\nabla \mu_{\rm B}}{k_{\rm B}T} = \frac{\Phi}{C_{\rm B}} \nabla C_{\rm B}$$

Quenching or irradiation driving force

$$\frac{\nabla \mu_{\rm V}}{k_{\rm B}T} = \frac{1}{C_{\rm V}} \nabla C_{\rm V} - \frac{1}{C_{\rm A}} \frac{\partial \ln C_{\rm V}^{\rm eq}}{\partial \ln C_{\rm A}} \nabla C_{\rm A}$$
$$\frac{\nabla \mu_{\rm I}}{k_{\rm B}T} = \frac{1}{C_{\rm I}} \nabla C_{\rm I} - \frac{1}{C_{\rm A}} \frac{\partial \ln C_{\rm I}^{\rm eq}}{\partial \ln C_{\rm A}} \nabla C_{\rm A}$$

<sup>\*</sup>M. Nastar and F. Soisson, a Review on RIS published in Comprehensive Nuclear Materials, Elsevier

# Cea Diffusion rate theory for Radiation Induced Seg.

$$\begin{cases} \frac{dC_{V}}{dt} = K_{0} - RC_{I}C_{V} - div(J_{V}) \\ \frac{dC_{I}}{dt} = K_{0} - RC_{I}C_{V} - div(J_{I}) \\ \frac{dC_{A}}{dt} = -div(J_{A}^{V} + J_{A}^{I}) \\ \frac{dC_{B}}{dt} = -div(J_{B}^{V} + J_{B}^{I}) \end{cases}$$

Steady state profile

$$\begin{cases} J_{A} = 0 \\ J_{B} = 0 \\ J_{V} = J_{I} \end{cases}$$

Wiedersich formulae

$$\frac{\nabla \mathsf{C}_{_{\mathsf{B}}}}{\nabla \mathsf{C}_{_{\mathsf{V}}}} \! \propto \! - \! \left( \frac{\mathsf{L}_{_{\mathsf{BV}}}}{\mathsf{L}_{_{\mathsf{AV}}}} \! - \! \frac{\mathsf{L}_{_{\mathsf{BI}}}}{\mathsf{L}_{_{\mathsf{AI}}}} \right)$$

 $J_{A}^{V} = -L_{AA}^{V}\nabla\mu_{A} - L_{AB}^{V}\nabla\mu_{B} - L_{AV}\nabla\mu_{V}$ 





### PHASE FIELD METHOD FOR SPINODAL DECOMPOSITION

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### SPINODAL REGIME: A MEAN FIELD CONCEPT



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### THE CAHN-HILLIARD LINEAR STABILITY ANALYSIS (1958-65)



EXPERIMENTAL KINETICS: S AND LOG(S)



The early stage effective diffusion coefficient is positive and not not varying so much with k

### **CLUSTER DYNAMICS**



### Microstructural evolution under irradiation

\*after slides of T. Jourdan

### Formation of cavities, second phases, dislocation loops



SA 316 irradiated in EBR-II to 1.9x10<sup>26</sup> n/m<sup>2</sup> at 580 °C

Rate theory:  

$$\begin{cases}
\frac{dC_{V}}{dt} = G - K_{iv}C_{i}C_{V} - \sum_{j}k_{j,v}^{2}D_{V}(C_{V} - C_{V,j}^{eq}) \\
\frac{dC_{i}}{dt} = G - K_{iv}C_{i}C_{V} - \sum_{j}k_{j,i}^{2}D_{i}(C_{i} - C_{i,j}^{eq})
\end{cases}$$

Ex: a divacancy  $V_2$  should be treated as a sink, in  $k_{V_2,V}^2$  but  $C_{V_2}$  can also evolve  $\rightarrow$  Simulate the whole clustering process:  $\frac{dC_V}{dt}, \frac{dC_{V_2}}{dt}, \frac{dC_{V_3}}{dt}, L \frac{dC_{V_n}}{dt}$  DE LA RECHERCHE À L'INDUSTRI

## **Cluster Dynamics**

\*E. Clouet, Modeling of Nucleation Processes, ASM handbook 22A (2009)

#### Master equation



Flux in terms of V mobility

> Absorption coefficients  $\beta_{n,V}$ :  $\beta_{n,V} = 4\Pi r_n \frac{D_V}{\Omega} C_V$ 

Asumption of a steady state profile:  $\Delta C_{\rm V}=0$ 

 $\succ$  Emission coefficients  $lpha_{\mathrm{n,V}}$  :

$$J_{n \to n+1} = 0 \quad \Longrightarrow \alpha_{n,V}^{eq} = f(\beta_n)$$

Asumption:  $\alpha_{n,V} = \alpha_{n,V}^{eq}$ 

$$J_{n \to n+1} = \beta_n C_n - \alpha_{n+1} C_{n+1}$$

$$\beta_{n,V} \qquad \alpha_{n,V}$$

$$V \qquad n \qquad V$$

# **Cluster migration of clusters V<sub>3</sub>O in Fe**







### -Driven systems:

Not only the path but also steady states depend on Lij

<u>-Phase Field method</u> What is the spinodal diffusion coefficient? Is the Cahn-Hilliard kinetic law valid for spinodal decomposition?

<u>-Cluster Dynamics</u> Needs for a statistical definition of atom-based mobilities, association and dissociation rates



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### L<sub>IJ</sub> FROM THE EQUILIBRIUM FLUCTUATIONS

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## Diffusion of a random walker (vacancy)

$$\frac{1}{0 \ 1 \ 2 \ \cdots \ m}$$
Probability of being at m after N jumps:  $P(m,N) = \left(\frac{1}{2}\right)^{N} \frac{N!}{m! \ (N-m)!}$ 
Continuous limit:  $P(m,N) \approx \sqrt{\frac{2}{\Pi N}} \exp \frac{-m^{2}}{2N}$  Einstein relation (1905):  
m/N<<1,  $D = \frac{\sqrt{2}}{2t}$ 
Einstein relation (1905):  
 $D = \frac{\sqrt{2}}{2t}$ 
Mean free path  $\left\langle x^{2} \right\rangle = \sum_{i=1}^{N} \left\langle x_{i}^{2} \right\rangle + 2\sum_{i=1}^{N-1} \sum_{j=i+1}^{N} \left\langle x_{i}x_{j} \right\rangle$ 
Diffusion coefficient of a random walker Jump frequency:  $W_{0}$ 
Coordination number:  $Z$ 
Averaged number of jumps:  $\left\langle n_{jp} \right\rangle = Z \ \Gamma t$ 
BCC cristal  $\begin{cases} l_{x} = a/2 \\ Z = 8 \end{cases}$ 
 $D_{x} = \frac{1}{2} 8W_{0} \frac{a^{2}}{4} = W_{0}a^{2}$ 



### Vacancy in pure metal A

$$D_{\rm V} = \frac{\left\langle n_{\rm jp}^{\rm V} \right\rangle l^2}{6t}$$

A chosen atom in pure metal  $A = \text{tracer } A^*$  (red atom)

$$D_{A^*} = C_V \frac{\left\langle \left(R^{A^*}\right)^2\right\rangle}{6t} = C_V \frac{\left\langle n_{jp}^{A^*}\right\rangle l^2}{6t} f_0 \quad \text{with} \quad f_0 = \frac{\left\langle \left(R^{A^*}\right)^2\right\rangle}{\left\langle n_{jp}^{A^*}\right\rangle l^2}$$

### Estimation of the self-diffusion factor

H: nbre of vacancy escape paths



## BCC dilute alloy: correlation factor

#### Body centered cubic cristal



Pure metal

$$f_0 = \frac{H}{2+H} \quad avec \quad H = Z - 1$$

First shell approximation of the correlation factor

 $H = 7W_3$  Number of exit ways to leave the first shell of B

$$f_B = \frac{7W_3}{2W_2 + 7W_3}$$

\*Manning (1964) Leclaire (1970) Serruys and Brebec (1982) Garnier et al., Phys. Rev. B (2014)





First shell approximation of the correlation factor

$$f_B = \frac{H}{2W_2 + H} = \frac{2W_1 + 7W_3}{2W_2 + 2W_1 + 7W_3}$$

How do we calculate the correlation factor? How do we calculate the whole Onsager matrix? Definition of the Lij in terms of displacement fluctuations?

\*A. B. Lidiard, Phil. Mag. 46,1218 (1955)

### Tracer diffusion coefficient from Einstein relation

\*A. Einstein, Ann. Phys. Lpz. 17, 549 (1905) \*A. D. Le Claire, Phil. Mag. 3,921 (1958)

$$D_{A^*} = \frac{\langle r^2 \rangle}{6t}$$
 r: displt of a tracer A\* in time t

# Phenomenological coefficients

Kubo's linear response theory applied to transport in alloys \*A. R. Allnatt, J. of Chem. Phys. 43, 1855 (1965):



 $L_{ij} = \frac{\langle R_i R_j \rangle}{6Vt}$   $R_i$ : collective displacement of species i in time t

Calculation methods based on fluctuations:

-Analytical approach (Bardeen and Herring, 1951; Manning 59, ; random lattice gas models (Manning 69), transition Matrix (Moleko)

-Computational approach: Atomic Kinetic Monte Carlo

### L<sub>IJ</sub> FROM DISSIPATION PROCESSES

First pair mean field calculation of the Lij in concentrated alloys \*R. Kikuchi, Ann. Phys. N. Y.II, 306 (1961); J. Phys. Chem. Solids 20,17,35 (1961).



# Point mean field fluxes



Alloy A,A\*,V

Microscopic flux of A\* between two sites

$$J_{A^{*}}^{i \to i+1} = \omega_{A^{*}} C_{A^{*}}^{i} C_{V}^{i+1} - \omega_{A^{*}} C_{A^{*}}^{i+1} C_{V}^{i} \simeq -\omega_{A^{*}} C_{V} d_{klm} \frac{C_{A^{*}}^{i+1} - C_{A^{*}}^{i}}{d_{klm}} \simeq -\omega_{A^{*}} C_{V} d_{klm} \nabla_{i} C_{A^{*}}$$

Macroscopic flux

$$J_{A^{*}} = \frac{Z_{klm}}{S_{klm}} J_{A^{*}}^{i \to i+1} = -\frac{Z_{klm} d_{klm}}{S_{klm}} \omega_{A^{*}} C_{V} \nabla_{i} C_{A^{*}} = -\frac{Z_{klm} d_{klm}^{2}}{S_{klm} d_{klm}} \omega_{A^{*}} C_{V} \nabla_{i} C_{A^{*}} = -\frac{a^{2}}{\Omega_{at}} \omega_{A^{*}} C_{V$$

Tracer diffusion coefficient:  $D_{A^*} = \omega_{A^*} a^2 C_V$ 

A point mean field approximation does not yield the correlations



## Pair probability: correlations

No pair probability: no correlation





Non equilibrium pair probabilities as state variables not deduced from local concentrations

# Non equilibrium distribution function

Occupation numbers

$$n = \left\{ n_i^{\alpha}, n_i^{\beta}, \cdots, n_j^{\alpha}, n_j^{\beta} \right\}$$

Equilibrium distribution function

$$P_0(n) = \exp\left[\left(\Omega + \sum_{\alpha} \mu_{\alpha} \sum_{i} n_i^{\alpha} - H(n)\right) / k_B T\right]$$

Non equilibrium distribution function

$$P(n,t) = P_0(n)P^{(1)}(n,t) \qquad n_j^{V}$$

$$P^{(1)}(n,t) = \exp\left[\left(\delta\Omega + \sum_{\alpha} \mu_{\alpha} \sum_{i} n_{i}^{\alpha} \delta\mu_{i}^{\alpha}(t) - h(n,t)\right) / k_{B}T\right]$$

Effective hamiltonian

$$h(n,t) = \frac{1}{2!} \sum_{i,j} \sum_{\alpha,\beta} v_{ij}^{\alpha\beta}(t) n_i^{\alpha} n_j^{\beta} + \frac{1}{3!} \sum_{i,j,k} \sum_{\alpha,\beta} v_{ijk}^{\alpha\beta\gamma}(t) n_i^{\alpha} n_j^{\beta} n_k^{\gamma} + \cdots$$

### What is the evolution of the effective interactions?





### Pair probability: correlations



### Moments of the probability

$$\left\langle n_{i}^{\alpha} \right\rangle = \sum_{n} n_{i}^{\alpha} P(n,t)$$
  
 $\left\langle n_{i}^{\alpha} n_{j}^{\beta} \right\rangle = \sum_{n} n_{i}^{\alpha} n_{j}^{\beta} P(n,t)$ 

### Equilibrium probability

$$\left\langle n_{i}^{\alpha} n_{j}^{\beta} \right\rangle^{(0)} = \sum_{n} n_{i}^{\alpha} n_{j}^{\beta} P_{0}(n) = \left\langle n_{i}^{\alpha} \right\rangle^{(0)} \left\langle n_{j}^{\beta} \right\rangle^{(0)} y_{\alpha\beta}$$
  
Dilute alloy  $y_{\alpha\beta} = e^{-\frac{E_{\alpha\beta}^{b}}{kT}} E_{\alpha\beta}^{b}$ : binding energy

Pair correlators in terms of the effective interactions

$$k_{ij}^{\alpha\beta} = \left\langle n_i^{\alpha} n_j^{\beta} \right\rangle - \left\langle n_i^{\alpha} \right\rangle \left\langle n_j^{\beta} \right\rangle$$
$$k_{ij}^{\alpha\beta} - k_{ij}^{\beta\alpha} = \left\langle n_i^{\alpha} n_j^{\beta} \right\rangle^{(0)} \left( v_{ij}^{\alpha\beta} - v_{ij}^{\beta\alpha} \right) = \left\langle n_i^{\alpha} n_j^{\beta} \right\rangle^{(0)} a_{ij}^{\alpha\beta}$$

Relationships between effective interactions due to the normalisation of pair probablities

$$\sum_{\beta} k_{ij}^{\alpha\beta} = 0, \quad \forall \alpha \quad \blacksquare \qquad \Rightarrow \qquad a_{ij}^{AV} = -\frac{C_B}{C_V} a_{ij}^{AB} = -\frac{C_B}{C_A} y_{BV} a_{ij}^{BV}$$
  
where  $y_{BV} = e^{-\frac{E_{BV}^b}{k_B T}}$ 

Low vacancy concentration:

$$C_V << C_A, C_B \Rightarrow a_{ij}^{AB} << a_{ij}^{BV}$$

$$C << C \Rightarrow a^{AV} << a^{BV}$$

Specific case of a dilute alloy A(B):  $C_B \ll C_A \implies a_{ij}^{AV} \ll a_{ij}^B$ 

In sums of anti-symmetric interactions, a single interaction to be considered:





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### Length of vacancy loops related to the range of the effective interactions

# SCMF microscopic flux

 $J_{i\to s}^{\alpha} = \delta_{is} \left\langle n_i^{\alpha} n_s^{V} W_{is}^{\alpha V} \left( \delta \mu_i^{\alpha} + \delta \mu_s^{V} - \sum_{j,\beta} v_{ij}^{\alpha \beta} n_j^{\beta} - \sum_{j,\beta} v_{sj}^{V\beta} n_j^{\beta} + v_{is}^{\alpha V} \right) \right\rangle^{\gamma}$ 

(0)

 $-\delta_{is}\left\langle n_{i}^{V}n_{s}^{\alpha}W_{is}^{\alpha V}\left(\delta\mu_{i}^{V}+\delta\mu_{s}^{\alpha}-\sum_{j,\beta}v_{sj}^{\alpha\beta}n_{j}^{\beta}-\sum_{j,\beta}v_{ij}^{V\beta}n_{j}^{\beta}+v_{is}^{V\alpha}\right)\right\rangle^{(0)}$ Deta

Detailed balance: 
$$\left\langle n_{i}^{\alpha} n_{s}^{V} W_{is}^{\alpha V} \right\rangle^{(0)} = \left\langle n_{i}^{V} n_{s}^{\alpha} W_{is}^{V \alpha} \right\rangle^{(0)}$$
  
Non equil. Chemical potentials:  $\delta \mu_{i}^{\alpha V} = \left( \delta \mu_{i}^{\alpha} - \sum_{j,\beta} v_{ij}^{\alpha \beta} n_{j}^{\beta} \right) - \left( \delta \mu_{i}^{V} - \sum_{j,\beta} v_{ij}^{V \beta} n_{j}^{\beta} \right)$ 

(0)

$$J_{i \to s}^{\alpha} = -\delta_{is} \left\langle n_{i}^{\alpha} n_{s}^{V} W_{is}^{\alpha V} \left( \delta \mu_{s}^{\alpha V} - \delta \mu_{i}^{\alpha V} \right) \right\rangle^{(0)} + \delta_{is} \left\langle n_{i}^{\alpha} n_{s}^{V} W_{is}^{\alpha V} \left( v_{is}^{\alpha V} - v_{is}^{V \alpha} \right) \right\rangle^{(0)}$$



# Flux of solute B in dilute alloys A(B)

$$J_{i \to s}^{B} = -L_{BB}^{(0)} \left( \delta \mu_{s}^{BV} - \delta \mu_{i}^{BV} \right) + L_{BB}^{(0)} \left( v_{is}^{BV} - v_{is}^{VB} \right)$$

### Non correlated Onsager coefficient

$$L_{BB}^{(0)} = \delta_{is} \left\langle n_i^B n_s^V W_{is}^{BV} \right\rangle^{(0)} = C_V C_B y_{BV} W_2$$





## Rate equations of pair correlations (1)

$$\begin{aligned} \frac{dk_{ij}^{BV}}{dt} &= \frac{d\left\langle n_{i}^{B} n_{j}^{V} \right\rangle - \left\langle n_{i}^{B} \right\rangle \left\langle n_{j}^{V} \right\rangle}{dt} \approx \frac{d\left\langle n_{i}^{B} n_{j}^{V} \right\rangle}{dt} - \left\langle n_{i}^{B} \right\rangle \frac{d\left\langle n_{j}^{V} \right\rangle}{dt} &= \delta_{ij} \left\langle n_{i}^{B} n_{j}^{V} W_{ij}^{BV} \left( \delta \mu_{j}^{BV} - \delta \mu_{i}^{BV} - a_{ij}^{BV} \right) \right\rangle^{(0)} \\ &+ \sum_{s \neq j} \delta_{js} \left\langle n_{i}^{B} n_{j}^{A} n_{s}^{V} W_{js}^{AV} \left( \delta \mu_{j}^{AV} - \delta \mu_{s}^{AV} \right) \right\rangle^{(0)} + A \\ &+ \left\langle n_{i}^{B} \right\rangle^{(0)} \sum_{s} \delta_{js} \left\langle n_{j}^{A} n_{s}^{V} W_{js}^{AV} \left( \delta \mu_{j}^{AV} - \delta \mu_{s}^{AV} - a_{ij}^{AV} \right) \right\rangle^{(0)} \\ &+ C_{B} C_{V} y_{BV} \left[ -4 W_{3} \left( \delta \mu_{i+2}^{AV} - \delta \mu_{i+1}^{AV} \right) + \left( 2 W_{1} + W_{3} \right) \left( \delta \mu_{i+1}^{AV} - \delta \mu_{i}^{AV} \right) \right] + A \\ &+ C_{B} C_{V} W_{0} \left[ 4 \left( \delta \mu_{i+2}^{AV} - 2 \delta \mu_{i+1}^{AV} + \delta \mu_{i}^{AV} \right) - 4 \left( a_{i+1,i+2}^{AV} - a_{i,i+1}^{AV} \right) \right] \end{aligned}$$



Now we need to calculate

# Rate equations of pair correlations (2)

$$A = \sum_{s \neq j} \delta_{js} \left\langle n_{i}^{B} n_{j}^{A} n_{s}^{V} W_{js}^{AV} \left( v_{is}^{BV} + v_{ij}^{BA} + v_{sj}^{VA} - v_{is}^{BA} - v_{ij}^{BV} - v_{sj}^{AV} \right) \right\rangle^{(0)}$$

$$B = -\sum_{s \neq j} \delta_{js} \left\langle n_{i}^{B} n_{j}^{A} n_{s}^{V} W_{js}^{AV} \right\rangle^{(0)} \left( v_{ij}^{BV} - v_{ij}^{BA} \right) = -C_{B} C_{V} y_{BV} (7W_{3} + 4W_{1}) \left( v_{i,i+1}^{BV} - v_{i,i+1}^{BA} \right)$$

$$C = \sum_{s \neq j} \delta_{js} \left\langle n_{i}^{B} n_{j}^{A} n_{s}^{V} W_{js}^{AV} \left( v_{is}^{BV} - v_{is}^{BA} - a_{sj}^{AV} \right) \right\rangle^{(0)} = C_{B} C_{V} y_{BV} \left[ 2W_{1} (v_{i,i+1}^{BV} - v_{i,i+1}^{BA}) - 2(W_{1} + W_{3}) a_{i,i+1}^{AV} + 4W_{3} a_{i+1,i+2}^{AV} + 2W_{1} (v_{i,i}^{BV} - v_{i,i}^{BA}) \right]$$

$$A = B + C = C_B C_V y_{BV} \left[ -(7W_3 + 2W_1)(v_{i,i+1}^{BV} - v_{i,i+1}^{BA}) - 2(W_1 + W_3)a_{i,i+1}^{AV} + 4W_3 a_{i+1,i+2}^{AV} + 2W_1(v_{i,i}^{BV} - v_{i,i}^{BA}) \right]$$



# Rate equations of antisymmetric pair interactions

$$\frac{1}{C_B C_V} \frac{dk_{ij}^{BV}}{dt} = y_{BV} W_2 \Big( \delta \mu_{i+1}^{BV} - \delta \mu_i^{BV} - a_{i,i+1}^{BV} \Big) + y_{BV} \Big[ -4W_3 \Big( \delta \mu_{i+2}^{AV} - \delta \mu_{i+1}^{AV} \Big) + (2W_1 + W_3) \Big( \delta \mu_{i+1}^{AV} - \delta \mu_i^{AV} \Big) \Big] \\ + W_0 \Big[ 4 \Big( \delta \mu_{i+2}^{AV} - 2\delta \mu_{i+1}^{AV} + \delta \mu_i^{AV} \Big) - 4 \Big( a_{i+1,i+2}^{AV} - a_{i,i+1}^{AV} \Big) \Big] \\ + y_{BV} \Big[ -(7W_3 + 2W_1) (v_{i,i+1}^{BV} - v_{i,i+1}^{BA}) - 2(W_1 + W_3) a_{i,i+1}^{AV} + 4W_3 a_{i+1,i+2}^{AV} + 2W_1 (v_{i,i}^{BV} - v_{i,i}^{BA}) \Big] \Big]$$



$$\frac{da_{i,i+1}^{BV}}{dt} = 2W_2 \Big( \delta\mu_{i+1}^{BV} - \delta\mu_i^{BV} - a_{i,i+1}^{BV} \Big) + \Big[ -4W_3 \Big( \delta\mu_{i+2}^{AV} - \delta\mu_i^{AV} \Big) + 2(2W_1 + W_3) \Big( \delta\mu_{i+1}^{AV} - \delta\mu_i^{AV} \Big) \Big] \\
+ \frac{W_0}{y_{BV}} \Big[ 4 \Big( \delta\mu_{i+2}^{AV} - 3\delta\mu_{i+1}^{AV} + 3\delta\mu_i^{AV} - \delta\mu_{i-1}^{AV} \Big) \Big] \\
+ \Big[ -(7W_3 + 2W_1)a_{i,i+1}^{BV} + 2W_1 (v_{i,i}^{BV} - v_{i+1,i+1}^{BV} - v_{i,i}^{BA} + v_{i+1,i+1}^{BA}) \Big]$$

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### The case of uniform gradients of chemical potentials

$$\begin{aligned} \frac{da_{i,i+1}^{BV}}{dt} &= 2W_2 \Big( \delta\mu_{i+1}^{BV} - \delta\mu_i^{BV} - a_{i,i+1}^{BV} \Big) + \Big[ -4W_3 \Big( \delta\mu_{i+2}^{AV} - \delta\mu_i^{AV} \Big) + 2(2W_1 + W_3) \Big( \delta\mu_{i+1}^{AV} - \delta\mu_i^{AV} \Big) \Big] \\ &+ \frac{W_0}{y_{BV}} \Big[ 4 \Big( \delta\mu_{i+2}^{AV} - 3\delta\mu_{i+1}^{AV} + 3\delta\mu_i^{AV} - \delta\mu_{i-1}^{AV} \Big) \Big] \\ &+ \Big[ -(7W_3 + 2W_1)a_{i,i+1}^{BV} + 2W_1 (v_{i,i}^{BV} - v_{i+1,i+1}^{BV} - v_{i,i}^{BA} + v_{i+1,i+1}^{BA}) \Big] \end{aligned}$$

Uniform gradients of chemical potentials:

$$\begin{cases} \nabla \mu_{\alpha V} = \delta \mu_{i+1}^{\alpha V} - \delta \mu_{i}^{\alpha V} = \delta \mu_{i+2}^{\alpha V} - \delta \mu_{i+1}^{\alpha V} = \delta \mu_{i}^{\alpha V} - \delta \mu_{i-1}^{\alpha V} \\ v_{i,i}^{\alpha \beta} = v_{i+1,i+1}^{\alpha \beta} \end{cases}$$

Adiabatic approximation on  $a_{i,i+1}^{BV}$  because characteristic time of  $a_{i,i+1}^{BV}$ 

is proportional to  $1/(7W_3 + 2W_2)$  while characteristic time of solute B is  $1/(C_VW_2)$ 

$$\frac{da_{ij}^{BV}}{dt} = 0 \quad \Longrightarrow \quad a_{i,i+1}^{BV} = \frac{2W_2 \nabla \mu_{BV} + 2(2W_1 - 3W_3) \nabla \mu_{AV}}{7W_3 + 2W_1 + 2W_2}$$

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# SCMF calculation of the Lij in FCC

$$a_{i,i+1}^{BV} = \frac{2W_2 \nabla \mu_{BV} + 2(2W_1 - 3W_3) \nabla \mu_{AV}}{7W_3 + 2W_1 + 2W_2}$$
$$J_{i \to s}^B = -y_{BV} C_B C_V W_2 \nabla \mu_{BV} + y_{BV} C_B C_V W_2 a_{i,i+1}^{BV} = -L_{BB} \nabla \mu_{BV} - L_{AB} \nabla \mu_{AV}$$

### Onsager coefficients Lij

$$\begin{cases} L_{BB} = y_{BV}C_BC_VW_2(1 - \frac{2W_2}{7W_3 + 2W_1 + 2W_2}) \\ L_{AB} = y_{BV}C_BC_VW_2\frac{2(3W_3 - 2W_1)}{7W_3 + 2W_1 + 2W_2} \end{cases}$$

Flux coupling under irradiation



### SCMF applied to dilute BCC and FCC alloys:

-In coll. with Univ. Illinois (D. Trinkle, P. Bellon): T. Garnier et al., PRB 88, 134108 (2013); PRB 88, 134201 (2013), PRB 89, 144202 (2014), PRB 90,184301 (2014) -In coll. with KTH (Pär Olsson and L. Messina): Messina et al.PRB 90, 104203 (2014) DE LA RECHERCHE À L'INDUSTRI

### General criteria for the solute drag

in coll. With Univ. Illinois, Garnier at al. PRB 88, 134201 (2013)



# 



| PAGE 71

E<sub>f</sub><sup>b</sup>



Mn.

### **Tracer diffusion coefficients**





[3] L. Messina et al., PRB 90, 104203 (2014) (includes also references to experimental values).




# Vacancy flux coupling in Fe(X)





| PAGE 74

# Vacancy-solute binding energies in Fe(X)



The magnitude and treshold temperature of solute drag is related to the solute-vacancy binding energy.

# Vacancy-direct interstitial flux coupling in Fe

### PhD of Thomas Schuler, SRMP, CEA-Saclay

Small solute concentrations

	X = C	X = N	X = O
max([X <sub>sol</sub> ]) (appm)	10 <sup>2</sup>	10 <sup>3</sup>	10 <sup>1</sup>

High vacancy-solute binding energy

	X = C	X = N	X = O
E <sup>bt</sup> (VX) (eV)	0.41	0.73	1.43
E <sup>bt</sup> (VX <sub>2</sub> ) (eV)	1.18	1.56	2.97

Transport coefficients (L<sub>ij</sub>): a macroscopic quantity

$$\begin{pmatrix} \vec{J}_V \\ \vec{J}_X \end{pmatrix} = - \begin{pmatrix} L_{VV} & L_{VX} \\ L_{XV} & L_{XX} \end{pmatrix} \begin{pmatrix} \vec{\nabla}\mu_V \\ \vec{\nabla}\mu_X \end{pmatrix}$$



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### Flux coupling including paires VX



\*Nastar, PRB 90, 144101 (2014)

$$J_{i \to s}^{B} = -y_{BV}C_{B}C_{V}W_{2}\nabla\mu_{BV} + y_{BV}C_{B}C_{V}W_{2}a_{i,i+1}^{BV}$$

$$\frac{da_{i,i+1}^{BV}}{dt} = 2W_2 \Big( \delta\mu_{i+1}^{BV} - \delta\mu_i^{BV} - a_{i,i+1}^{BV} \Big) + \Big[ -4W_3 \Big( \delta\mu_{i+2}^{AV} - \delta\mu_i^{AV} \Big) + 2(2W_1 + W_3) \Big( \delta\mu_{i+1}^{AV} - \delta\mu_i^{AV} \Big) \Big] \\
+ \frac{W_0}{y_{BV}} \Big[ 4 \Big( \delta\mu_{i+2}^{AV} - 3\delta\mu_{i+1}^{AV} + 3\delta\mu_i^{AV} - \delta\mu_{i-1}^{AV} \Big) \Big] \\
+ \Big[ -(7W_3 + 2W_1)a_{i,i+1}^{BV} + 2W_1 (v_{i,i}^{BV} - v_{i+1,i+1}^{BV} - v_{i,i}^{BA} + v_{i+1,i+1}^{BA}) \Big]$$

Adiabatic approximation on  $a_{i,i+1}^{BV}$  because the characteristic time of  $a_{i,i+1}^{BV}$ 

is proportional to  $1/(7W_3 + 2W_2)$  while characteristic time of solute B is  $1/(C_V W_2)$ 

$$\frac{da_{i,i+1}^{BV}}{dt} = 0 \quad \Rightarrow J_B = -L_{BB}\nabla\mu_{BV} - L_{AB}\nabla\mu_{AV} - Q_{BB}\nabla^3\mu_{BV} - Q_{AB}\nabla^3\mu_{AV} - \cdots$$

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# Vacancy kinetic correlation effects driven by non uniform driving forces





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### Interdiffusion in binary alloys with non uniform driving forces

#### Non homogeneous linear response theory

$$\int_{B} = -L_{BB} \nabla \mu_{BV} - L_{AB} \nabla \mu_{AV} - Q_{BB} \nabla^{3} \mu_{BV} - Q_{AB} \nabla^{3} \mu_{AV}$$
$$\frac{dC_{B}}{dt} = -\nabla J_{B} = L_{BB} \nabla^{2} \mu_{BV} + L_{AB} \nabla^{2} \mu_{AV} + Q_{BB} \nabla^{4} \mu_{BV} + Q_{AB} \nabla^{4} \mu_{AV}$$

#### nonconservative vacancy

conservative vacancy

$$D = C_A D_B + C_B D_A$$

$$D_B = \left(\frac{L_{BB} - k^2 Q_{BB}}{C_B} - \frac{L_{AB} - k^2 Q_{AB}}{C_A}\right) (C_A C_B \Phi_{CH})$$

$$D_A = \left(\frac{L_{AA} - k^2 Q_{AA}}{C_A} - \frac{L_{AB} - k^2 Q_{AB}}{C_B}\right) (C_A C_B \Phi_{CH})$$

$$D_B = \left(\frac{L_{BB}}{C_B} - \frac{L_{AB}}{C_A}\right) C_A C_B \left[\frac{\partial^2 f}{\partial C_B^2} + (\kappa_E + \kappa_C)k^2\right]$$

$$D_A = \left(\frac{L_{AA}}{C_A} - \frac{L_{AB}}{C_B}\right) C_A C_B \left[\frac{\partial^2 f}{\partial C_B^2} + (\kappa_E + \kappa_C)k^2\right]$$

$$D = \frac{\left(L_{AA} - k^2 Q_{AA}\right) \left(L_{BB} - k^2 Q_{BB}\right) - \left(L_{AB} - k^2 Q_{AB}\right)^2}{L_{AA} + L_{BB} + 2L_{AB} - k^2 \left(Q_{AA} + Q_{BB} + 2L_{AB}\right)} \Phi_{CH}$$

$$D \simeq \frac{L_{AA}L_{BB} - L_{AB}^{2}}{L_{AA} + L_{BB} + 2L_{AB}} \left[ \frac{\partial^{2} f}{\partial C_{B}^{2}} + (\kappa_{E} + \kappa_{C})k^{2} \right]$$



SCMF in satisfying agreement with Monte Carlo: the deviation from Cahn-Hilliard comes from the heterogeneous correlation effects DE LA RECHERCHE À L'INDUSTRI

### Effect of the correlations on D(k<sup>2</sup>)

T/Tc=2.4 and  $C_B=0.1$  $\kappa_E > 0$  $\kappa_c < 0$  $v_{\rm B} = 10 v_{\rm A}$  $V_{\rm A} = V_{\rm B}$ Cahn-Hilliard 1.6 Cahn-Hilliard Monte Carlo Monte Carlo 5 SCMF 1.4 4 1.2-Δ Δ 3 1.0 0.8  $\Lambda^{10}$ 5 15 Ó  $\Lambda^{10}$ 15 5 0  $D_{SCMF} = M \left[ \frac{\partial^2 f}{\partial C_{P}^2} + \left( \kappa_E + \kappa_C \right) \Lambda_k + \kappa^{nl} {\Lambda_k}^2 \right]$ 

SCMF prediction of a possible change of sign of the composition gradient parameter due to correlations is verified



# CONCLUSION

### Diffusion

A multi-scale phenomenon that can be characterized at various scales (from nm to microns).

However there is a crucial lack of diffusion data: diffusion experiments do not catch the large variety of point defect behaviors, they are mostly measured at high Temperature.

### **SCMF** theory

An atomic diffusion theory that integrates kinetic correlations produced by a point defect diffusion mechanism.

It gives a clear definition of the mesoscopic and macroscopic thermodynamic and kinetic parameters and in this way facilitates a step by step multiscale approach.

Moreover, it yields new kinetic laws challenging the phenomenological models such as the Cahn-Hillard method.



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