$\begin{array}{c} \textbf{Structure de la surface (010) de Al}_{13}\textbf{Fe}_{4} \\ \\ \textbf{Émilie Gaudry} \end{array}$

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Institut Jean Lamour - UMR7198

- IJL, UMR 7198 CNRS Université de Lorraine
- \simeq 450 personnes
- dont \simeq 180 chercheurs / enseignants chercheurs
- 4 départements:
 - Physique de la Matière et des Matériaux
- Chimie et Physique des Solides et des Surfaces
- Science et Ingénierie des Matériaux et Métallurgie
- Nanomatériaux, électronique et vivant









Équipe Métallurgie et Surfaces

- 5 permanents: Jean-Marie Dubois, Vincent Fournée, Julian Ledieu, Marie-Cécile de Weerd, Émilie Gaudry.
- Alliages métalliques complexes: volume, interfaces, surfaces, films minces
- Moyens: Cluster de calcul de l'IJL / IDRIS + plateformes expérimentales





Introduction

Structure de la surface (010) de $AI_{13}Fe_4$,

approximant des quasicristaux décagonaux (d-QC)

- Réactivité chimique: semi-hydrogénation de l'acétylène
- Interaction de la structure 3D du système massif avec la structure 2D de surface



Une description détaillée de la structure de surface est nécessaire pour la compréhension des propriétés de surface

Introduction

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Structure of the decagonal approximant $Al_{13}Fe_4$

- \checkmark Crystal structure C2/m(mC102): a=15.492 Å, b=8.078 Å, c=12.471 Å, $eta=107.69(1)^0$
- Stacking of flat (17 AI + 8 Fe) and puckered (22 AI + 4 Fe) atomic layers along the [010] direction





Structure of the decagonal approximant $Al_{13}Fe_4$

High symmetry atomic clusters (PB) as structural building blocks





PB = 23 atom cluster (16 Al + 7 Co) Columns of PB altern. with junction layer Fe-Al-Fe molecule aligned along [010] Grin *et al.*, *Z. Kristallogr.*, **209** (1994) 479 Grin *et al.*, CMA, Wiley (2010).

Al₁₃Fe₄: Experimental Results



- Single crystal by the Czochralski method (P. Gille et al, Crystal research and technology, 43 (2008) 1161) Surface preparation: cycles of Ar⁺ bombardment / annealing
- P No surface reconstruction (From LEED $\frac{a}{c} = 1.24$, in agree. with Grin et al., Z. Kristallogr., **209** (1994)
- Angle-dependent XPS (not shown here) demonstrates that surface segregation can be disregarded.
 - Step height measurements ($\simeq b/2$) highlight a surface plane selection



Al₁₃**Fe**₄: **Surface Plane Selection**



Step height measurements ($\simeq b/2$)highlight a surface plane selectionHigh corrugationSurface atomic densitySurface composition

DFT calculations (PAW-PBE)

Calc.	Exp.		
<i>a</i> = 15.43 Å	<i>a</i> = 15.49 Å		
<i>b</i> = 8.03 Å	<i>b</i> = 8.08 Å		
<i>c</i> = 12.43 Å	<i>c</i> =12.47 Å		
β = 104 0	eta = 107 0		

8-layers asymetric slabs

Tersoff-Hamann approximation

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8-layers asymetric slabs (\simeq 200 at.)

Tersoff-Hamann approximation

Al₁₃**Fe**₄: **Surface Plane Selection**



High Resolution STM



Al₁₃**Fe**₄: **Surface Electronic Structure**





 \checkmark High corrugated surface (partial exposition of the F subsurface layer)

DOS dominated by the Fe d-states

Consistency between the experimental data and the calculated DOS

Conclusion

Surface plane selection: Surfaces form at specific bulk planes charact. by a high atomic density and a high content of the lowest surf. energy element / Consistent with the "plane selection rules" for QCs

Interplay of the 3D cluster structure with the 2D surface:

Compound	Surface	Pure Bulk trunc.	3D struct. preserv.	Ref.
$AI_{13}Fe_4$	built from the P layer	no	yes	PRL
	High corrugated surface			(2013)
o -Al $_{13}$ Co $_4$	Built from the P layer	almost yes	no	PRB 84 (2011)
(Czochralski)	Pure Al plane as surf. term.	without surf. Co at.		085411

o-Al₁₃Co₄ $Pmn2_1$ (oP102)



Reactivity: active sites might be the isolated Fe atoms protruding above pentagonal motifs

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Role of the chemical bonding: $E_{strength}^{AlTM} = E_{f}^{lac-AlTM} - (E_{f}^{lac-Al} + E_{f}^{lac-TM})$ $E_{strength}^{AlCo} = -0.31eV$ $E_{strength}^{AlFe} = -0.66eV$



Reactivity: active sites might be the isolated Fe atoms protruding above pentagonal motifs

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Thank you for your attention!



